

# Mass Spectrometric Detection of Homocysteine, Methylmalonic Acid and Succinic Acid using HILIC on a Zwitterionic Stationary Phase.

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## Introduction

Rapid, simple, rugged yet sensitive are prerequisites often sought for analytical techniques used in clinical laboratories. The vast amount of samples in difficult matrices, (i.e., urine, plasma, serum etc.) requires techniques on which uncomplicated methods can be developed and used, preferably unmanned, with little or no downtime. A typical example is the quantification of homocysteine, methylmalonic acid and succinic acid; Figure 1. Homocysteine and methylmalonic acid (MMA) are diagnostic markers for B12-deficiency while succinic acid is a physiologically abundant isomer to MMA which often interferes with the quantification of the former compound. Over the years, yet in vain, different approaches using various analytical techniques have been tested with the overall goal to allow quantification of all three compounds within the same run.

Today there is a feasible alternative, hydrophilic interaction liquid chromatography (HILIC) combined with mass spectrometric detection (MS). HILIC is a separation technique suitable for polar and hydrophilic compounds, it uses an eluent containing a high content of water miscible organic solvent (e.g., acetonitrile) to promote hydrophilic interactions between the analyte and a hydrophilic stationary phase. Although there are quite a few HILIC phases commercially available, none is truly comparable with the ZIC®-HILIC/pHILIC phases. The highly polar zwitterionic columns provide a unique environment particularly capable of solvating polar and charged compounds, which enables high performance HILIC separations. The zwitterionic stationary phase, Figure 2, can interact with charged analytes via weak electrostatic interactions, and in practice, this provides the chromatographer with a larger degree of freedom when choosing among buffer salts and ionic strength in method development, thus making the column an ideal choice for LC-MS analysis. This particular application note illustrates advantages when combining efficient separation with a sensitive detection principle, and is exemplified by an isocratic HILIC separation of homocysteine, methylmalonic acid and succinic acid.

## ZIC®-HILIC/pHILIC Columns

The ZIC®-HILIC columns are silica based stationary phases with either 3.5, 5 or 10 µm particle size, while the ZIC®-pHILIC columns are polymer-based stationary phases with 5 µm particle size, yet both having a sulfobetaine type zwitterionic functionality.

## Experimental Conditions

Column: ZIC®-HILIC 50 × 4.6 mm, 5 µm

### UV

Column temp: RT  
 Mobile phase: Acetonitrile/ammonium acetate (100 mM, pH 6.8); 70/30 (v/v)  
 Flow-rate: 1.5 mL/min  
 Detector: UV at 206 nm (UFS 1.0 V)  
 Injection volume: 5 µL of test solution in mobile phase

### MS

Column temp: 30 °C  
 Mobile Phase: Acetonitrile/ammonium acetate (100 mM, pH 6.8); 75/25 (v/v)  
 Flow-rate: 1.0 mL/min  
 Split: 100 µL/min to MS  
 Detector: MS, ESI in positive mode  
 Capillary voltage: 3000 V  
 Fragmentor: 150 V  
 Mass range: 50–200 m/z  
 Injection volume: 5 µL of 0.1 mg/mL of each compound diluted in mobile phase  
 Sample: In elution order; homocysteine, methylmalonic acid and succinic acid all dissolved in mobile phase.

Figure 1: Structures of (1) homocysteine, (2) methylmalonic acid and (3) succinic acid.

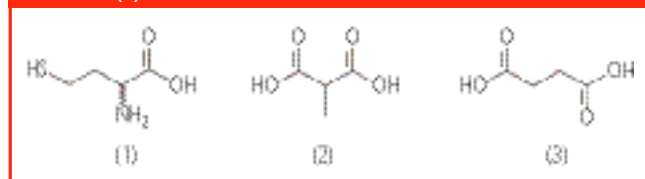
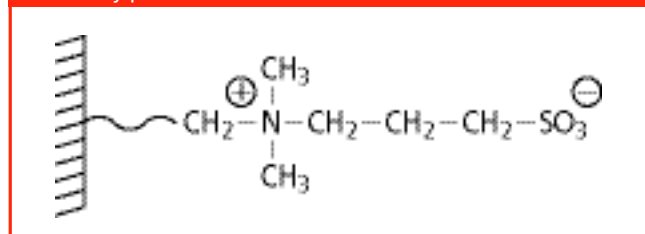


Figure 2: Schematic representation of the ZIC®-HILIC stationary phase.



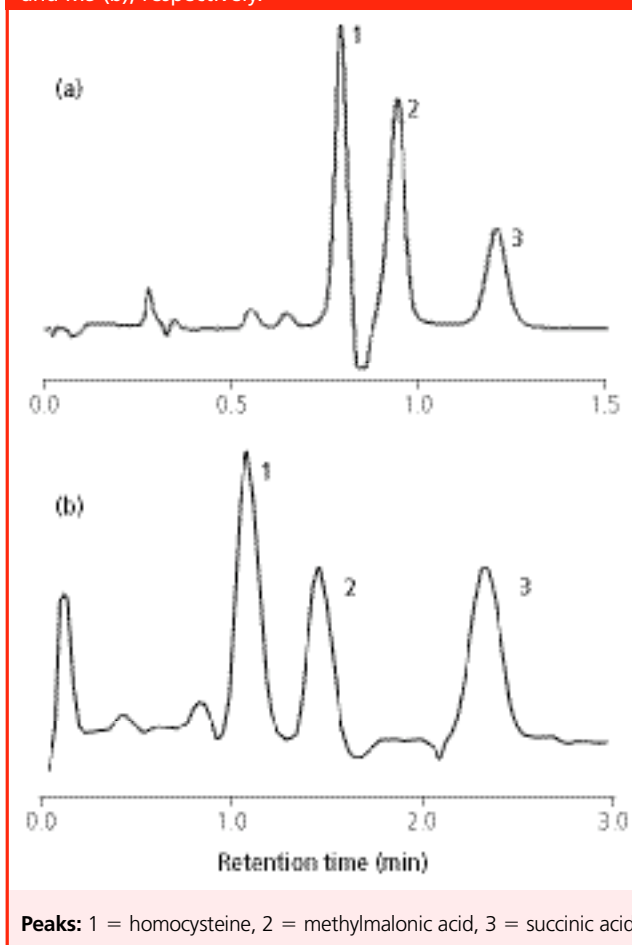
## Results

Method development is commonly performed using UV detection, because of its ease of use and robustness, but the technique often lacks the sensitivity needed to allow quantification at relevant physiological levels. Herein, it is illustrated that provisional optimal separation conditions can be established via UV-detection, Figure 3(a), and then easily transferred and slightly modified to better fit MS detection in order to gain sensitivity. Baseline separation for all compounds can be achieved within 90 seconds, and that the compounds elute with a  $k'$  between 1.4 and 3. Worth noting is the dip in between homocysteine and methylmalonic acid. The rationale for the phenomena is a combination of low detection wavelength and a slight mismatch in buffer concentration between the sample and the mobile phase. When transferring the separation from the LC–UV to the LC–MS instrumental set-up, it became evident that the ionic strength was compromising the detection, and that a higher efficiency was needed in order to compensate for additional extra-column effects between the column and the detector. Lowering the flow-rate and the aqueous portion in the mobile phase (from 30 to 25% volume), and slightly decreasing the ionic strength, sufficient separation efficiency was reached, as seen in Figure 3(b). An overall ionic strength of 25 mM ammonium acetate is typically too high for optimized MS sensitivity, yet for the particular application it is a necessity from a chromatographic perspective. Using the provisional MS compatible conditions, clinical relevant concentration levels can easily be quantified, however by switching to selected ion monitoring and optimizing typical mass spectrometer parameters such as capillary voltage, nebulizer gas flow, drying gas pressure and drying gas temperature, the sensitivity may improve by a factor of 10–50 depending on the fragmentation.

## Conclusion

The ZIC®–HILIC column is indeed a suitable tool for separation of homocysteine, methylmalonic acid and succinic acid. Combined with MS detection, physiological relevant concentration can easily be quantified with the possibility of processing up to 20 samples per hour.

**Figure 3:** Chromatograms of the separation of homocysteine, methylmalonic acid and succinic acid, using detection by UV (a) and MS (b), respectively.



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